

References

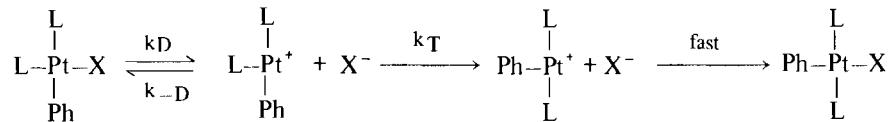
- 1 H. Huchital and R. J. Hoages, *Inorg. Chem.*, **12**, 998 (1973).
- 2 J. Casabó and J. M. Coronas, *An. Quim.*, **70**, 330 (1974); J. Casabó, J. M. Coronas and J. Ribas, *Inorg. Chim. Acta*, **11**, 149 (1974); J. Casabó, J. Ribas and J. M. Coronas, *J. Inorg. Nucl. Chem.*, in press; J. Casabó, J. M. Coronas and M. Ferrer, *Inorg. Chim. Acta*, in press.
- 3 J. P. Birk, *J. Am. Chem. Soc.*, **91**, 3189 (1973).
- 4 D. W. Hoppenjans and J. B. Hunt, *Inorg. Chem.*, **8**, 506 (1969).
- 5 E. Kyuno and M. Kamada, *Bull. Chem. Soc. Japan*, **40**, 1849 (1967).
- 6 N. Nakamoto, "Infrared spectra of Inorganic and Coordination Compounds", 2nd ed., Wiley, New York (1970).
- 7 D. F. Shriver, *Struc. Bonding*, **1**, (1966); H. J. Coerver and C. Curran, *J. Am. Chem. Soc.*, **80**, 3522 (1958).
- 8 Landolt-Börstein, "Zahlenwerte und Funktionen aus Physik-Chemie", 7 Teil, Berlin (1966).
- 9 H. J. King, *J. Chem. Soc.*, **127**, 2100 (1925).

ERRATA to Volumes 14 and 15

Factors Affecting Dissociative and Associative Mechanisms on Platinum(II) Complexes

R. ROMEO, D. MINNITI and M. TROZZI, *I4*, L15 (1975):

The reaction scheme should read:



The Structure of Pd^{II} Complexes Containing the Tridentate Ligand Diethylenetriamine and its N-alkyl Derivatives. Part I. Molecular Structure of Nitrodiethylenetriaminepalladium(II) Nitrate and Nitro-1,1,7,7-tetraethyl-diethylenetriaminepalladium(II) Nitrate

N. BRESCIANI, M. CALLIGARIS, L. RANDACCIO, V. RICEVUTO and U. BELLUCO, *I4*, L17 (1975):

Line 9 from the bottom on page L18, left column should read:

"the normal orientation of the NO₂ group is preferred because of the steric repulsion between the NO₂ group and the"

Copper(II) Complexes of α -Oximinoketones

J. C. DANILEWICZ, R. D. GILLARD and R. WOOTTON, *I5*, L5 (1975):

In the text to structure I on page L5, the correct version reads: (la) R₁ = Ph, R₂ = CH₃

Preparation and Doping of CaLaZrTaO₇ Pyrochlore

S. LARACH, *I5*, L8 (1975):

The top paragraph, right column, should read:

"It is of interest that the material formed by cold-pressing contained two pyrochlore phases, in addition to ZrO₂. The major pyrochlore phase was the same as the one reported in the hot-pressing case, with a lattice constant of 10.625 Å. The second pyrochlore phase had a lattice constant of 10.74 Å, and was possibly LaTaO₄."